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(54) Premix compositions for use in the production of flame retardant polyurethane foams.

(57) The storage stability of otherwise unstable premix compositions for flame retardant polyurethane foams containing in the premix a halogenated polyol and a tertiary amine catalyst can be enhanced by incorporating into the premix a condensation product of acrylic or methacrylic acid with an alcohol or an amine.

This invention relates to premix compositions for use in the production of flame retardant rigid polyurethane foams and is particularly concerned with improving the storage stability of such premix compositions.

Premix compositions for rigid flame retardant polyurethane foams containing halogenated polyols are unstable at room or elevated temperature when such premix compositions contain a tertiary amine catalyst. Apparently decomposition of the halogenated polyol gives rise to HX which forms a tertiary amine hydrohalide salt and deactivates the catalyst.

We have now found that the stability of such premix compositions can be enhanced by inclusion therein of a condensation product of acrylic or methacrylic acid with an alcohol or amine.

Among the esters and amides of acrylic and methacrylic acid found particularly beneficial are those corresponding to the formulae:

$$H_2C = C - C - N$$

$$R_2$$

$$R_2$$
(1)

$$H_2C = C - C - O (C_nH_{2n}) Y$$
 (II)

$$(H_2C = C - C - O CH_2)_r - C - B_t$$
 (III)

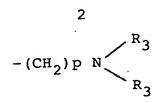
wherein R is H or CH₃; R₁ is H, an alkyl radical of 1 to 4 carbon atoms, or a hydroxyalkyl radical of 2 to 3 carbon atoms; R₂ is an alkyl radical of 1 to 4 carbon atoms, a hydroxy alkyl radical of 2 to 3 carbon atoms, or a substituted amino alkyl group of the formula

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5 wherein each R_3 is independently an alkyl radical of 1 to 4 carbon atoms or a hydrogen alkyl radical of 2 to 3 carbon atoms, and p is 2 to 4;

Y is H, OH, N or
$$-0 - \frac{0}{C} - \frac{R}{C} = CH_2$$

wherein each R_4 is independently an alkyl radical of 1 to 4 carbon atoms and n is 1 to 12, provided that n is 2 when Y is OH or $N \subset \mathbb{R}^4$; and

B is hydrogen or an alkyl radical of 2 to 4 carbon atoms, r is 3 or 4, t being zero when r is 4 and t being 1 when r is 3.

particularly preferred condensation products comprise dimethylamino ethyl acrylate, dimethylamino ethyl methacrylate, isobutyl methacrylate, and mixtures thereof.

Advantageously the tertiary amine catalyst is dimethylethanolamine.

Typically, the premix composition will also include a blowing agent and a surfactant cell stabilizer. The blowing agent typically comprises a chlorofluorocarbon and the cell stabilizer is typically a siloxane compound.

In use, the required di- or poly- isocyanate is added to the premix composition to initiate the isocyanate-polyol reaction and blowing to produce the desired rigid flame retardant cellular polyurethane.

The premix composition may contain any of the known tertiart amine catalysts employed in the production of rigid cellular polyurethanes or mixtures of such catalysts. Particular examples of these include triethylene diamine, dimethyl ethanolamine, dimethyl cyclohexylamine, triethylam-

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ine, N-methyl morpholine, dimethyl piperazine, tetramethyl propane diamine. More frequently employed are the aliphatic tertiary amines, particularly dimethylethanolamine.

The surfactant employed as a cell stabilizer may be of the silicone oil type such as polysiloxane-polyoxyalkylene block copolymers or polydimethyl siloxane; or the type obtained by polymerization of dibutyl maleate monomer and an N-vinyl pyrrolidone monomer compound in a polyol polymerization medium (U.S. Patent No. 3,746,663).

Any of the usual blowing agents generally employed in rigid polyurethane formulations may be employed, preferably those of the halohydrocarbon type, such as trichlorofluoromethane.

The halogenated polyol may be a halogenated aliphatic or cycloaliphatic polyester or polyether polyol or mixtures thereof having 2 to 8 hydroxyl groups, there being at least one chlorine or bromine atom attached to a carbon of the polyol. Typical chlorinated polyols more frequently employed in flame retardant polyurethanes are the chlorinated polyether polyols having hydroxyl number values of from about 200 to about 700 and having a chlorine content in the range of about 25 to 60% by weight. Particularly preferred is a chlorinated polyether polyol having a hydroxyl number of 365[±] 10 and containing about 47% by weight of chemically bound 25 chlorine.

A particularly preferred premix composition comprises, by weight

halogenated polyether polyol 100 chlorofluorocarbon blowing agent 45 dimethylethanolamine 3.5 condensation product 2 to 4

To such premix composition may be added 20 parts, by weight, of an amino polyol for reducing surface friability of polyurethane foam, said amino polyol having a hydroxyl number of $600^{\frac{1}{2}}$ 10 and a viscosity of 280 cps at 25° C, and optionally a cell stabilizing siloxane compound.

EXAMPLE 1

A series of experimental runs were made to determine the effectiveness of the condensation product in retarding deactivation of the catalyst in the premix for fire retardant rigid polyurethane foams. In all of these runs the same basic premix composition was employed. The basic premix composition comprised:

			. bpm
- 0		(Thermolin TM RF-230 ⁽¹⁾	100
10		Thermolin TM RF-230 ⁽¹⁾ Poly G^R , 70-600 ⁽²⁾	20
	Basic	Genetron ^R , RllSBA ⁽³⁾	. 45
	Premix:	Cell stabilizer (4)	1.5
		Tertiary amine catalyst	(as indicated)
15		Condensation product	(as indicated)

- (1) A reactive chlorinated polyether polyol supplied by Olin Corporation Designed Products Division for use in production of flame retardant rigid polyurethanes. The product contains 47% by weight chemically bound chlorine and has a hydroxyl number of 365[±] 10 (mg KOH/gm) and a viscosity of 100,000 cps at 25°C.
- (2) An amino-polyol supplied by the above-named Olin Division specifically designed to reduce surface friability in rigid polyurethane foams based on Thermolin RF230. It has a hydroxyl number of 600[±] 10 (mg KOH/gm) and a viscosity of 280 cps at 25^oC.
 - (3) Trichloro monofluoromethane blowing agent.
- (4) LK-221TM or DC-193 copolymers were employed.

 LK-221 copolymer, supplied by Air Products and Chemicals, Inc., is a copolymer of N-vinyl pyrrolidone and dibutyl maleate polymerized in a trifunctional polyol. DC-193, supplied by Dow-Corning comprises polysiloxane-polyoxyalkylene block copolymers as described in US Patents 2,834,748 and 2,917,480.

Before and after storage of the foregoing premix composition at about $38-41^{\circ}C$ for indicated time periods, the activity was determined by reaction with Mondur R MR 5 .

(5) A polymethylene polyphenol isocyanate supplied by Mobay Chemical Co., containing 32% NCO groups, having an equivalent weight of 132.

The results are reported in Tables 1 and 2 comparing the stabilized premix compositions with an unstabilized premix composition, each employing dimethyl ethanol amine (DMEA) catalyst (3.5 parts catalyst in stabilized compositions and 3.7 parts in the control).

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Hydroxyethyl methacrylate	ण	14	40	45	80	17	21	15	10	
rox)	ام	13	38	42	75	0	15	8	m	
Hyd met	αi	12	33	39	73	•	ı	1	1	
.Hydroxyethyl methacrylate 3.0	٦	13	37	47	. 62	c o	ε0	27	ស	
coxye 18Cry 3.0	ച	13	37	43	76	8	80	16	7	
Hyd)	. ស	12	34	37	75	1	ı	ı	ı	
	ળ	14	39	46	80	17	15	24	77	
ethy col cryl 2.0	ام.	13	37	42	81	80	6	14	13	
Triethylene glycol diacrylate 2.0	100	12	34	37	72	1	1	t	1	
	OI	13	38	48	86	18	12	17	10	
ethyl ane cryle 2.0	ام	13	36	42	78	18	9	7	0	•
Trimethylol propane triacrylate	៧	11	34	41	78	ı	1		, 1	
amino ylate	U	11	30	32	89	. 0	0	0	80	
hyl acr	ما	. 11	29	33	. 65	0	ó	0	က	
Dimethyl amino ethyl acrylate	, as		30	34	63	1	1	٠,	1	
ate te	υ	13	. 38	45	89	18	12	15	0	
ryl cryl	ما	13	37	47	. 60	18	σ	21		
n-Lauryl Methacrylate	, a	. I	34	39	78	1	1	ı	1	
ĽΣ	4	<u>=</u> 19	49	71	90	78	58	.65	73	
巨	o o	" 1	26	49	11	22	16	17	20	
NONE	α	11 O	31	43	59		•	ı		
Condensation Product (Stabilizer)	мдд	in the Caca .	Cream crime (secs.)	Tack-free time	<u></u>	Percent loss Cream time	Gel time	Tack-free time	Rise time	

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Condensation						TABLE	2					
Product (Stabilizer) pbw		None	_	Ia Meth	Isobutyl thacryla	Isobutyl Methacrylate 3.0	Is Met	Isobutyl ethacryl	Isobutyl Methacrylate 4.0	1,3 dim	-But ethy 2.0	l,3-Butylene dimethyacrylate 2.0
	ल।	ខា	441	ıαl	QΙ	ום.	छ।	ᆈ	ण	ια	ام	 (10)
Cream time (secs.)	6	11	16	13	13	14	13	13	14	12	13	12.
<pre>Gel time</pre>	31	26	49	34	37	37	34	37	36	33	35	37
Tack-free time (secs.	43	49	71	40	44	46	39	43	45	37	43	45
Rise time (secs.)	59	71	06	75	82	81	78	71	19	79	75	75
Percent loss Cream time	i m o	22	78		0	Φ,	i	0.	8	ŧ	æ	0
Gel time	•	16	58	1	6	க்		6	9	ı	9	12
Tack-free time	ŧ	17	65	ı	10	15°	. 1	10	15	ı	,16	
Rise time	1	20	73	ŧ	6	8	ı	0	-	1	0	0

a) initial
b) after 11 days
c) after 19 days
d) after 18 days
e) after 8 days
f) after 15 days
g) after 15 days

g () () () () () () ()

	3. (cry	υI	14	38	44	75	ω	12	13	15
	Isobutyl methacry- lamide 3.0	ы	13	36	43	74	0	9	10	14
•	I m I m	ø۱	13	34	39	65	1	i	1	1
	Dimethylamino- ethyl methac- rylate 2.0	υl		30	33	62	6	7	0	0
	imethylami thyl metha rylate 2.0	ы					0	7	က	0
	imet thy]		11	30	35	09				
	О́в	σl	11	28	34	65	ı	,	·	1 .
gg)	Dimethylamino- propyl meth- acrylamide 2.0	ত।	12	29	36	57	6	7	70	2
TABLE 2 (Continued)	ethyle opyl n crylan 2.0	ام	12	29	34	58	6	7	13	4
	Dime pro ac	٠ ,	11	27	30	26	i	ı	ŧ	ı
	ane	च।	14	38	43	76	17	12	16	4
	l,6-Hexane diol diacrylate 2.0	ام	13	36	44	78	8	9	19	7
	1, 6 dia	σı	12	34	37	73	1	i	•	1
•	Trimethylol Propane trimethac- rylate 2.0	ξ	13	37	40	11.	8	12	80	0
	imethylo Propane imethac	.al	13	35	44	71	æ	9	19	0
	444	ત્યા	12	33	37	74	1 10	t	1	1
	Condensation Product (Stabilizer) pbw		Cream time (secs.)	<pre>Gel time (secs.)</pre>	Tack-free time (secs.)	Rise time (secs.)	Percent loss Cream time	Gel time	Tack-free time	Rinse time

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The low activity losses indicate that the premix compositions stabilized with the condensation products of Tables 1 and 2 can be successfully used after storage at ambient temperature for prolonged periods of up to about six months.

In addition to the condensation products disclosed in Tables 1 and 2, other examples of compounds corresponding to Formula (I) include:

bishydroxyethylacrylamide dibutyl acrylamide diethyl acrylamide

Examples of compounds corresponding to Formula (II) in addition to those specifically named in the Tables above, are:

hydroxyethyl acrylate hydroxyethyl methacrylate

An example of a compound corresponding to Formula 20 (III) in which r=4, is

pentaerythritol tetraacrylate

The rapid loss in activity (over 50% loss) shown by the unstabilized control is a common occurrence not only with the DMEA catalyst used in the reported test but also is found with other tertiary amine catalysts when employed in premix compositions containing halogenated polyols. The condensation products of the present invention are effective in retarding the rate of loss in activity of fire retardant polyurethane premix compositions employing such other tertiary amine catalysts.

Results of stability tests made on flame retardant polyurethane premix compositions with other tertiary amine 35 catalysts are reported in Tables 3 and 4.

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TABLE 3

STABILIZER		NONE		•	DMAEM 1.0	—	מ	DMAEM 3.0		מ	DMAEM 1.0	
CATALYST		DMEA			DMEA 3.5		A	DMEA 3.5			(a) 2.5	
(Mad)	A	ပ	[FJ.	A	В	О	A	В	Ω	Æ	æ	Ω
Cream time (secs.)	6	11	16	10	11	13	10	10	10	11	13	14
Gel time (Secs.) 31	ë.	. 56	49	32	34	34	30	29	30	30	34	38
Tack-free time	43	. 49	11	38	38		34	32	37	36	42	47
Rise time (secs.)	59	71	06	72	74	75	61	68	65	64	73	77
Percent loss Cream time	1	22	78	ı	10	30	1	0	0	1	18	27
Gel time	1	16	28	1	9	9		0	0	1	13	27
Tack-free	1	18	65	1	0	13	1	0	က	1	17	31
Rise time	ı	20	73	ľ	က	4	t	11	17.	1	14	70

initial || || || || ||

8 days after 7 days

after 14 days after 15 days after after O D E

= dimethylaminoethyl methacrylate DMAEM

DMEA = dimethyl ethanolamine - 80% dimethylethanolamine (a) = 20% triethylene diamine - 67% dipropylene glycol (b) = 33% triethylene diamine - 67% dipropylene glycol

TABLE 3 (Continued)

STABILIZER DMAEM (pbw) 5.0 cATALYST (a) (pbw) A B D A Cream time 10 10 13 13 (secs.) 27 27 29 33 Tack-free time 31 33 33 39 (secs.) (secs.) Rise time 53 61 55 60 (secs.)	DMAEM 1.0 (b) 3.5 B B 14 1 A 45 5 45 5 70 7	. A B B B B B B B B B B B B B B B B B B	3.0 3.5 B 3.5 B 3.5 3.5 3.5 3.5 3.5 3.5	4. 70	1 1 2 2	DMAEM 5.0 (b) 3.5 B B 28 34 35	D D 133 337 337
3 17 0 4	D D 29 33 55 6	(b) 3.5 3.5 13 13 14 1 29 33 36 4 33 39 45 5 55 60 70 7	1.0 (b) 3.5 D A B D A 13 13 14 18 13 29 33 36 43 28 33 39 45 50 35 55 60 70 77 65	1.0 3.0 (b) (b) (b) 3.5 3.5 3.5 13 13 14 18 13 14 29 33 36 43 28 32 33 39 45 50 35 37 55 60 70 77 65 62	1.0 3.0 (b) (b) 3.5 3.5 3.5 D A B D A B D 13 13 14 18 13 14 14 29 33 36 43 28 32 35 33 39 45 50 35 37 43 55 60 70 77 65 62 66	(b) 3.5 D A B D A B D A 13 13 14 18 13 14 14 12 29 33 36 43 28 32 35 27 33 39 45 50 35 37 43 33 55 60 70 77 65 62 66 48	(b) (b) (b) (b) (c) (d) (d) (d) (e) (

A = initial B = after 7 days D = after 14 days

DMAEM = dimethylaminoethyl methacrylate (a) = 20% triethylene diamine - 80% dimethylethanolamine (b) = 33% triethylene diamine - 67% dipropylene glycol

4	I
ы	l
2	

		D 24 25 29 24 4 4 4 4 4 24 24
5.0	30	110 23 22 28 28 28 28 28 28 28 28 28 28 28 28 28 2
E S	72	110 23 25
		D 29 33 54 54 17 17
3.0	2.0	B 26 31 51 10 10 11 11 11 11
		A 10 24 46 46 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
		D 31 37 64 64 20 24 23 36
DMAEM 1.0	2.0 2.0	B 12 28 39 39 55 55 12 13 17
Ω		10 10 255 330 477 471 111 111 1111 1111 1111 1111 11
_		10 D 288 33 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
S.0	(c) 4.5	B 112 239 333 57 60 0
: :		A 12 30 30 31 61
æ		112 339 339 0 0 2
DMAEM 3.0	(c)	112 33 33 33 66 66 0 0
•		A. 112 34 34 53 54 53 54 54 54 54 54 54 54 54 54 54 54 54 54
		D
MAEM 1.0	(c)	B 1336 444 73 0 0
ב		13 37 71 71 71 1 1 1 1 1 1 1 1 1 1 1 1 1
TABILIZER	(pow) ATALYST (pbw)	Cream time (secs.) Gel time (secs.) Tack-free (secs.) Rise time (secs.) Percent loss Cream time Gel time Tack-free Rise time
STABILIZER	(pbw) CATALYST (pbw)	Cream time (secs.) Gel time (secs.) Tack-free (secs.) Rise time (secs.) Percent loss Cream time Gel time

(c) = dimethylaminoethyl mofpholine

(d) = C-methyl triethylene diamine

In preparation of the activity-stabilized premix compositions of the invention only the usual amounts of tertiary amine catalyst need be employed as in the known formulations for rigid cellular polyurethane products. The total amount of catalyst plus condensation product will in general comprise 5.5 to 7.5 parts per hundred parts of halogenated polyol in the premix composition.

CLAIMS

- 1. A premix composition for use in the production of flame retardant polyurethane foams and comprising a tertiary amine catalyst and a halogenated polyol characterized in that said premix composition also contains a condensation product of acrylic or methacrylic acid with an alcohol or amine to enhance the storage stability of said premix composition.
- A premix composition according to Claim 1, characterized in that said condensation product is of the formula from 10 the group consisting of

$$H_{2}C = C - C - N$$

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$$H_2C = C - C - O (C_nH_{2n})Y$$

 $(H_2C = C - C - OCH_2)_r - C - B_t$

wherein R is H or CH₃; R₁ is H, an alkyl radical of 1 to 4 carbon atoms, a hydroxyalkyl radical of 2 to 3 carbon atoms;

R₂ is an alkyl radical of 1 to 4 carbon atoms, a hydroxy alkyl radical of 2 to 3 carbon atoms; or a substituted amino alkyl group of the formula

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$$-(CH_2)p - N < R_3$$

wherein each ${\bf R}_3$ is independently an alkyl radical of 1 to 4 carbon atoms, or a hydroxy alkyl radical of 2 to 3 carbon atoms, and p is 2 to 4

Y is H, OH,
$$\stackrel{R}{N}_{4}$$
 or $-0-\stackrel{O}{C}-\stackrel{R}{C}=CH_{2}$

10 wherein each R_4 is independently an alkyl radical of 1 to 4 carbon atoms, and n is 1 to 12; provided that n is 2 when Y is OH or $N \underbrace{ R_4^{4}}_{R_4}$; and

B is hydrogen or an alkyl radical of 2 to 4 carbon atoms, r is 3 to 4, t being zero when r is 4 and t being l when r is 3.

- 3. A premix composition according to Claim 1, characterized in that said condensation product is dimethylamino ethyl acrylate.
- A premix composition according to Claim 1, character ized in that said condensation product is dimethylamino ethyl methacrylate.
 - 5. A premix composition according to Claim 1, characterized in that said condensation product is isobutyl methacrylate.
- 25 6. A premix composition according to any preceding Claim, characterized in that said tertiary amine catalyst is dimethylethanolamine.
 - 7. A premix composition according to any preceding Claim, characterized in that said halogenated polyol is a chlorin-
- 30 ated polyether polyol having a hydroxyl number of $365^{\frac{1}{2}}$ 10 and containing about 47% by weight of chemically bound chlorine.
 - 8. A premix composition according to any preceding Claim, characterized in that it comprises by weight:

halogenated polyether polyol

chlorofluorocarbon blowing agent

dimethylethanolamine

condensation product

100

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3.5

2 to 4

9. A premix composition according to Claim 8, character5 ized in that it further comprises 20 parts, by weight, of an amino polyol for reducing surface friability of polyurethane foam, said amino polyol having a hydroxyl number of 600[±] 10 and a viscosity of 280 cps at 25°C.

10. A premix composition according to Claim 8 or 9, 10 characterized in that it further comprises a cell-stabilizing siloxane compound.

For the Applicants

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CHARTERED PATENT AGENT EUROPEAN PATENT ATTORNEY

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EUROPEAN SEARCH REPORT

Application number

EP 80 30 2164.1

		ERED TO BE RELEVANT		CLASSIFICATION OF THE APPLICATION (Int. CL3)
Category	Citation of document with indicat passages	ion, where appropriate, of relevant	Relevant to claim	
				C 08 G 18/18
İ				C 08 G 18/50
A	DE - A1 - 2 822 81	9 (OLIN CORP.)	1,6,	C 08 L 71/02
	* claims 1, 12; pa	ge 9, lines 19 to 26;	7, 10	C 08 K 5/20
	page 14, lines 7	to 17 *		C 08 K 5/17
				C 08 K 5/10
A	US - A - 3 980 579	(SYROP et al.)	1,6,	
	* claims 1,11; col	umn 3, lines 32 to	7,10	
	42; column 5, li	nes 48 to 55 *		
A	US - A - 3 248 369	(WEIDMAN et al.)	1,3,4	TECHNICAL FIELDS SEARCHED (Int.CL3)
	* claim 1; column	4, lines 3 to 11 *		
	•			C 08 G 18/00
				C 08 K 5/00
				C 08 L 71/00
				C 08 L 75/00
				CATEGORY OF
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				Lincitation for other reasons
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X	The present search report t	nas been drawn up for all claims		family, corresponding document
lace of sea	arch Dat	e of completion of the search	Examiner	201000010119
PO Form 1	Berlin	01-10-1980		MARX